

The Micro-canonical Algorithm

- The preceding discussion leads to the algorithm
- We have a system (e.g. N two state atoms) with total energy E (e.g. $E = N\Delta\bar{n}$) and want the temperature
- We "just" need to count the number of ways $\Omega(E)$ for the system to share (or partition) the energy. This determines the entropy $S(E)$:

$$S(E) = k_B \ln \Omega(E)$$

Then

$$\frac{\partial S(E)}{\partial E} = \frac{1}{T}$$

This determines the relation between the temperature and energy $E(T)$.

Example of Microcanonical Algorithm: Two State System

• You will do this in Homework. So I will only sketch the steps.

• Take N two level atoms with N_1 of them excited; $E = N_1 \Delta = N \bar{n} \Delta$ with $\bar{n} = N_1 / N$. The number in the ground state is $N_0 = N - N_1$, or $N_0 = N(1 - \bar{n})$. \bar{n} is the mean number of quanta of energy per site $\bar{n} < 1$. We found

$$\begin{aligned} \frac{S}{k_B} &= \ln \Omega = -N_0 \ln \frac{N_0}{N} - N_1 \ln \frac{N_1}{N} \\ &= N \left[-(1 - \bar{n}) \ln(1 - \bar{n}) - \bar{n} \ln \bar{n} \right] \end{aligned}$$

• As the energy increases $E = N \Delta \bar{n}$ increases

• Then Differentiating we can express \bar{n} in terms of T

$$\frac{1}{T} = \frac{\partial S}{\partial E} = \frac{1}{N \Delta} \frac{\partial}{\partial \bar{n}} N k_B \left[-(1 - \bar{n}) \ln(1 - \bar{n}) - \bar{n} \ln \bar{n} \right]$$

$$\frac{1}{k_B T} = \frac{1}{\Delta} \ln \left(\frac{1 - \bar{n}}{\bar{n}} \right)$$


do it in HW!

• For $\bar{n} = 1/4$ we find $k_B T = \Delta / \ln 3$

- We can now solve for \bar{n} (in Homework)

$$\bar{n} = \frac{1}{e^{\Delta/kT} + 1} = \frac{e^{-\Delta/kT}}{1 + e^{-\Delta/kT}}$$

- This agrees with the canonical approach (i.e. Partition Fcn)

$$\bar{n} = \frac{N_1}{N} = \text{probability to be in excited state}$$


Previously we found

$$P_1 = \frac{e^{-\Delta/kT}}{Z_1} = \frac{e^{-\Delta/kT}}{1 + e^{-\Delta/kT}}$$

Comments

- Clearly the canonical approach (the partition fcn way) is easier. But we didn't derive it. We relied on an intuitive notion of temperature, and postulated the probability $P_r \propto e^{-\epsilon_r/kT} =$
- The temperature is just a parameter in the Boltzmann Probability P_r , which is adjusted to reproduce the energy of the system.
- In the next section we will derive the Boltzmann factor from the $S(E)$