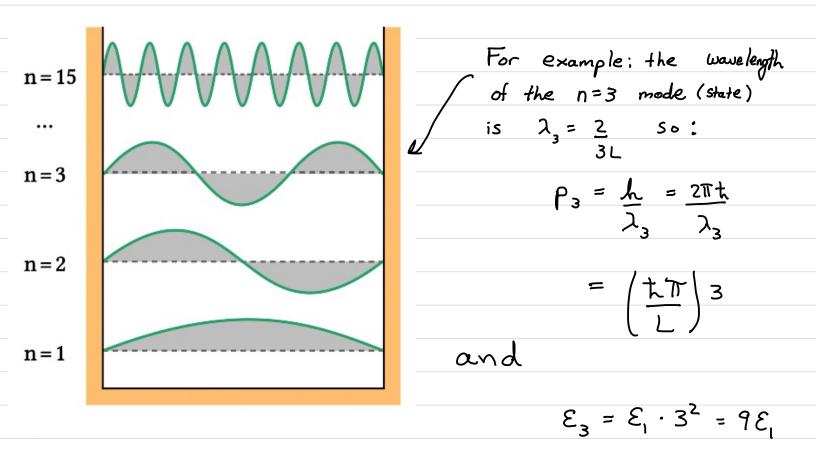
Particle in a Box

We will now explore the quantum mechanical "particle in the Box". This will justify a result I quoted earlier  $\sum \longrightarrow \int \frac{dx \, dp}{h}$ i.e. that the sum over states becomes an integral over classical phase space The Energy Levels and eigen-functions are:  $\mathcal{E}_{n} = \frac{\hbar^{2} \pi^{2} n^{2}}{2mL^{2}} \equiv \frac{p_{n}^{2}}{2m} \quad \text{with} \quad P_{n} \equiv \left(\frac{\hbar}{L}\right)^{n}$ this is the  $= \mathcal{E}_{1} n^{2} \text{ with } \mathcal{E}_{1} = \frac{t^{2} \pi^{2}}{2ml^{2}}$ magnitude of the 2m12 momentum of the with n=1,2,3,... n-th state wave functions are The  $\mathcal{U}_{n} = \sqrt{\frac{2}{L}} \sin\left(\frac{p_{n} \times}{t}\right) \iff \text{These are shown} \\
 below$ ~ e<sup>ipn×/t</sup> - e<sup>-ipn×/t</sup> each box state is a superposition of a right moving wave (+) and a left moving wave (-)

The partice in box wavefons



The quantum number n counts how many half-wavelengths fit in the box. For a typical box L~Im and typical atom λ-1Å~10<sup>-10</sup> m, n is huge, n~10<sup>10</sup>!
 So n (which labels the momentum pn=ħπn/L) is

practically continuous except at low temperatures of boxes of order an atomic length.

Particle - In - Box ; Stat Mech  $Z = \sum_{n=1}^{\infty} e^{-\beta \varepsilon_n} = \sum_{n=1}^{\infty} e^{-\beta \varepsilon_n n^2}$ Now the partition function can't be evaluated in closed form (at this level). But, we know that n is nearly continuous and very large. We can replace the sum with an integral.  $\sum_{n=1}^{\infty} \longrightarrow \int_{0}^{\infty} dn = \int \frac{L \, dp}{\pi \, t} = \int \frac{L \, dp}{2\pi \, t}$ one is we used  $P = \frac{k\pi}{L}n$  Then instead of integrating very small n~10'0 over the momentum magnitude momentum itsself, p = -00...00 inserting a factor of 2. So we see that for  $n \gg 1$ 

 $\sum_{n} \longrightarrow \int \frac{dx \, dp}{(2\pi t)} = \int \frac{dx \, dp}{h}$ 

So then the quantum treatment just reproduces the  
classical result in this limit:  

$$\begin{aligned}
& \mathcal{E}_{n} = -\frac{p_{n}^{2}}{2m} \Rightarrow \frac{p^{2}}{2m} \\
& \mathcal{E}_{n} = -\frac{p_{n}^{2}}{2m} \\
& \mathcal{E}_{n} = -\frac{$$

Jummar small T BE >>1  $e^{-\beta \varepsilon_1} + e^{-4\beta \varepsilon_1}$  $\sum_{k=0}^{\infty} e^{-\beta \epsilon_{k} n^{2}}$  $\frac{\sqrt{2r}}{2} \left(\frac{kT}{\epsilon_1}\right)^{1/2} \text{ large } T$   $\frac{\sqrt{2r}}{2} \left(\frac{kT}{\epsilon_1}\right)^{1/2} \text{ classical approx}$ 2 11 L At small temperatures we can approximate Z by just including the first two terms in the sum. At high T the classical approximation is good. Then we compute:  $\mathcal{U} = \langle \varepsilon \rangle = - \frac{2 \ln 2}{2}$ This is shown below with the low and high T limits: 7 60 exact exact 6 classical approx two state approx:  $\varepsilon_1$ ,  $\varepsilon_2$ 50 classical approx 5 40 4 13/1 U/ɛ₁ 30 good at high З 20 2 at low 10 1 0 0 2 4 6 8 10 10 20 30 40 50 60 70 80 100 0 90 kT/ε<sub>1</sub> kT/ε<sub>1</sub>

## More Dimensions

The classical approximation 
$$\sum_{shete} \rightarrow \int d^{3} \times d^{3}p / d^{3}$$
 works  
in more dimensions  
In 3D the PIB energy levels are  
 $\sum_{n \times n_{y}n_{z}} = \frac{t^{2} T^{2} (n_{x}^{2} + n_{y}^{2} + n_{z}^{2}) = p_{nx}^{2} + p_{ny}^{2} + p_{nz}^{2}}{2m 2m 2m 2m}$   
with  $n_{x}=1,...00$  and similarly for  $n_{y}$  and  $n_{z}$ . For each  
direction we define a momentum component:  
 $p_{nx} \equiv t_{z}T^{2} \cdot n_{x} = magnitude of momentum in the
L x direction, with similar notation
in y and z directions.
The sum over states is
 $\sum_{nx} = \sum_{nx} \sum_{n=1}^{\infty} \sum_{n_{x}=1}^{\infty} \sum_{n_{x}=1}^{\infty} \int_{0}^{\infty} dn_{x} \int_{0}^{\infty} dn_{z}$   
 $= \int_{0}^{\infty} \frac{Ldp_{x}}{T + x} \int_{0}^{\infty} \frac{Ldp_{z}}{T + x}$   
 $= \int_{0}^{\infty} dx dp_{x} \int_{(2T + x)}^{\infty} \int_{2T + x}^{\infty} \int_{(2T + x)}^{\infty} d^{2}z dp_{z} = \int_{0}^{d^{2}z} d^{2}p (2T + x)^{2}$$